# ORGANIC CHEMISTRY

## Paper Code: CY

### Test Type: Test Series

**Duration: 2:30 Hours** 

# **CHEMISTRY-CY**

TEST SERIES GATE 2019

**BOOKLET SERIES B** 

Date: 13-01-2019 Maximum Marks: 100

#### **Read the following instructions carefully:**

- 1. Attempt all questions.
- 2. This question paper consists of 2 sections, General Aptitude (GA) for 15 marks and the subject specific GATE paper for 85 marks. Both these sections are compulsory. The GA section consists of 10 questions. Question numbers 1 to 5 are of 1-mark each, while question numbers 6 to 10 are of 2-mark each. The subject specific GATE paper section consists of 55 questions, out of which question numbers 11 to 35 are of 1-mark each, while question numbers 36 to 65 are of 2-mark each.
- 3. The question paper may consist of questions of **multiple choice type** (MCQ) and **numerical answer type**.
- 4. Multiple choice type questions will have four choices against (a), (b), (c), (d), out of which only **ONE** is the correct answer.
- 5. For numerical answer type questions, each question will have a numerical answer and there will not be any choices.
- 6. All questions that are not attempted will result in zero marks. However, wrong answers for multiple choice type questions (MCQ) will result in **NEGATIVE** marks. For all MCQ questions a wrong answer will result in deduction of <sup>1</sup>/<sub>3</sub> marks for a **1-mark** question and <sup>2</sup>/<sub>3</sub> marks for a **2-mark** question.
- 7. There is NO NEGATIVE MARKING for questions of NUMERICALANSWER TYPE.
- 8. Non-programmable type Calculator is allowed

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	Q.1-Q. 5 carry O	NE mark each.						
1.	In the following question, out of the four alternatives, select the alternative which best expresses the meaning of the idiom/phrase. To kick the habit							
	<ul><li>(a) To start a healthy practice</li><li>(c) To overcome an addiction</li></ul>		(b) To make a h (d) To have the	<ul><li>(b) To make a habit of hurting other's feelings</li><li>(d) To have the habit of overcoming obstructions.</li></ul>				
2.	In the following question, out of the four alternatives, select the alternative which best expresses the meaning of the idiom/phrase. <b>To zip it</b> (a) To move along very fast (b) Send a parcel by post (c) A rude way of telling someone to stop talking (d) To put something precious in a safe place							
3.	What should be the of radius 6m ? (Pro and the cylinder in	at should be the minimum height of water in a cylindrical glass to completely submerge a spherical iron b adius 6m ? (Provided the ball fits completely in the glass, i.e., difference between the radius of the sphe the cylinder in negligible).						
	(a) 9 m	(b) 4 m	(c) 8 m	(d) 7 m				
4.	Given below are two which one logically STATEMENTS : I. All A are B II. Some A are C (a) Only I follows (c) Neither 1 nor 2	wo statements follow follows. 2 follows	ed by two conclusions. A CONCLUSIONS 1. Some B are C 2. Some B are A (b) Only II follow (d) Both 1 and 2 f	ssuming the statements to the true, decide				
5.	Four students A, B, C, D of read in four different classes from class (v) to class (viii) and wears uniform of four different colours i.e., red, white, blue, yellow. The one who wears white shirt reads in class (vi). B wears red shirt and does not read in class (vii) or class (viii). C does not wear yellow colour shirt and D does not read in class (viii). Based on the information provided find out who reads in class (vi).(a) B(b) A(c) D(d) C							
	Q.6-Q. 10 carry TWO marks each.							
6.	A sentence has been given with a blank to be filled with an appropriate word. Choose the correct alternative. They were because all their plans had gone away. (a) Defeated (b) Rejected (c) Elated (d) Distraught							
7.	A sentence has bee	en given with a blank t	o be filled with an approp	riate word. Choose the correct alternative.				
	This brings nothing	g elsejoy to u	s.	(d) to				
8.	A person can complete a job in 120 days. On day 1 he starts the work alone. On day 2, he is assisted by another person who also completes the job in 120 days. On day 3, they are joined by another person who also completes the work in 120 days like this, everyday a new person with the same efficiency joins the work. In how many days the job will be completed							
	(a) 20	(b) 30	(c) 10	(d) 15				
9.	Six friends P, Q, R T, but to the left of (a) P	, S, T, U are sitting in S. U is not at the right (b) R	a row facing the East. R i at end. Who is at the extrem (c) Q	s between P and T. Q is just to the right of me right end ? (d) S				
10.	Find out what should come in place of question mark (?) ? 3, 7,23, ? 479							
	(a) 63	(b) 95	(c) 120	(d) 98				

(2)



#### Q.11-Q.35 carry one mark each.







Absolute R/S configuration at the centre A and B, are respectively 15.

(b) S, S





(a) R, R

16. The C–H bond with least amount of bond dissociation energy for homolytical cleavage among the following marked bonds is



19. The correct structure of flipping of the given compound (X) is





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- 20. Treatment of the hexapeptide Ala-Vla-Phe-Lys-Asp-Ala in separate experiments with the enzyms amino peptidase, chymotrypsin and pepsin respectively, gives
  (a) Ala + Val-Phe-Lys-Asp-Ala; Ala-Val-Phe + Lys-Asp-Ala; Ala-Val-Phe-Lys + Asp-Ala
  (b) Ala + Val-Phe-Lys-Asp-Ala; Ala-Val + Phe-Lys-Asp-Ala; Ala-Val-Phe-Lys + Asp-Ala
  (c) Ala + Val-Phe-Lys-Asp-Ala; Ala-Val-Phe + Lys-Asp-Ala; Ala-Val-Phe-Lys-Asp + Ala
  (d) Ala-Val-Phe-Lys-Asp + Ala; Ala-Val + Phe-Lys-Asp-Ala; Ala-Val-Phe-Lys + Asp-Ala
- 21. Among the following, the number of compounds which can participates as 'Diene' component in a Diels-Alder reaction is



22. The total number of compounds that give precipitate readily on reaction with  $AgClO_4$  is



23. The isoelectric point of the given compound will be

Ф. СООН <sup>(4.0)</sup> (9.47) СООН (2.0)

(corresponding pK values are mentioned in bracket). (Upto one decimal place).



The major product (P) formed in the following reaction, 24.





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28.



- 29. The reaction between terminal alkyne and alkyl azide in presence of  $CuSO_4$  (10 mol%) and sodium ascorbate gives product (P). In the product (P) the number of nitrogen atoms present is/are \_\_\_\_\_
- 30. The major product (P) formed in the following reaction,







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SPh

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- 37. A solution of 2.0 g of (+)–glyceraldehyde in 10.0 ml of water was placed in a 10 cm polarimeter tube. Using the sodium D line, a rotation of +1.47° was observed at 25°C. The specific rotation of (+)–glyceraldehyde is \_\_\_\_\_\_(upto one decimal place).
- 38. Consider the following reaction,



Total number of secondary –OH groups in product (B) is/are

- 39. Calculate  $[\alpha]_D$  of a 1M solution of 2-chloropentane in a 10 cm cell, when the observed rotation is +3.64°. (Rounded up to one decimal places).
- 40. Addition of  $Br_2$  to (Z)-2-butene gives
  - (a) (2R, 3R) dibromobutane through achiral brominium ion
  - (b) (2S, 3S) dibromobutane through achiral brominium ion
  - (c) Mixture of (2R, 3R) and (2S, 3S) dibromobutane through chiral brominium ion.
  - (d) Mixture of (2R, 3R) and (2S, 3S) dibromobutane through achiral brominium ion.
- 41. The number of optically active stereoiosmers possible for the following compound is  $H_3C$ —CH(OH)—CH(Br)—CH(OH)—CH<sub>3</sub>
- 42. The major product (P) formed in the following reaction sequence.

 $(\mathbf{X})$ 





43. The major product (P) formed in the following reaction,



44. The correct major (A) + minor (B) products with their relationship formed (respectively) in the following reaction



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 $\boxed{10}$ 







12



(c)



51. The major product (P) is

<sup>″/</sup>СН<sub>3</sub>



(d)

Ē H₃

53. In the following reaction sequence the number of oxygen atoms in the product (P) is/are





54. In the following reaction, the product (P) shows following spectral data

$$\int_{+}^{OSiMe_{3}} \int_{+}^{TiCl_{4}} (P)$$
M.F. = C<sub>8</sub>H<sub>14</sub>O<sub>3</sub>  
 $v_{max}$  (cm<sup>-1</sup>) = 1745, 1730  
 $\delta_{c}$  (*ppm*) = 202,176, 62, 48, 34, 22, 15  
 $\delta_{H}$  (*ppm*) = 1.21(6H, s), 1.8(2H, t, J = 7 Hz), 2.24(2H, t, J = 7Hz), 4.3(3H, s), 10.01(1H, s))  
The correct structure of product (P) is  
Me Me Me (b) Me Me



55. The number of <sup>13</sup>C NMR signals exhibited by the compounds (A-C), are, respectively



57. The major product (B) formed in the following reaction sequence,





56.

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 $\left(14\right)$ 



58. The major product (P) formed in the following reaction



59. Total number of pairs with correct order of the respective property is



60. The products A and B in the following reaction sequence are







61. A tetra peptide made up of natural amino acids, has alanine as the N-terminal residue which is coupled to a chiral amino acid. Upon complete hydrolysis, the tetrapeptide gives glycine, alanine, phenylalanine and luecine. The number of possible sequence of tetrapeptide is \_\_\_\_\_\_

Li 62.

The product 'P' is: (a) Non-aromatic (b) Aromatic

(c) Antiaromatic

(d) Homoaromatic



63. In the following compounds, incorrect order of their stability is





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65. Precocene is a compound that cause insect Larvae to pupate and can also be found in some plants (*Ageratum spp.*), where it may act as an insecticide. It was isolated in minute amounts and has the following spectroscopic details.

Mass spectrum :

m/z (high resolution gives  $C_{13}H_{16}O_3$ ). M-15(100%) and M-30 (weak)

<sup>1</sup>H NMR =  $\delta_H(ppm)$  = 1.34(6*H*,*s*), 3.80(3*H*,*s*), 3.82(3*H*,*s*), 5.54(1*H*, *d*, *J* = 10 *Hz*) 6.37(1*H*, *d*, *J* = 10 *Hz*), 6.42(1*H*,*s*), 6.58(1*H*,*s*)

The correct structure of Precocene is



\*\*\*\*\*\*\*



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#### Space for rough work





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ANSWER KEY									
PART-A									
<b>1.</b> (c)	<b>2.</b> (c)	<b>3.</b> (b)	4.	(d)	<b>5.</b> (b)				
<b>6.</b> (d)	<b>7.</b> (c)	<b>8.</b> (d)	9.	(d)	<b>10.</b> (b)				
	PART	'-B							
<b>11</b> . (d)	<b>12</b> . (b)	<b>13</b> . (b)	14.	(3)	<b>15</b> . (c)				
<b>16.</b> (d)	<b>12.</b> (d)	<b>18.</b> (b)	19.	(5) (b)	<b>20.</b> (a)				
<b>21.</b> (7)	22. (6)	<b>23.</b> (3)	24.	(b)	25. (c)				
<b>26.</b> (b)	27. (a)	<b>28.</b> (a)	29.	(3)	<b>30.</b> (b)				
<b>31.</b> (b)	<b>32.</b> (c)	<b>33.</b> (a)	34.	(58)	<b>35.</b> (c)				
<b>36.</b> (b)	<b>37.</b> (8.5 to 8.9)	<b>38.</b> (2)	39.	(34.0 to 3	4.5) <b>40.</b> (d)				
<b>41.</b> (2)	<b>42.</b> (b)	<b>43.</b> (d)	44.	(d)	<b>45.</b> (c)				
<b>46.</b> (c)	<b>47.</b> (a)	<b>48.</b> (a)	49.	(b)	<b>50.</b> (d)				
<b>51.</b> (a)	<b>52.</b> (b)	<b>53.</b> (4)	54.	(b)	<b>55.</b> (d)				
<b>56.</b> (d)	<b>57.</b> (b)	<b>58.</b> (c)	59.	(4)	<b>60.</b> (d)				
<b>61.</b> (4)	62. (b) EER E	<b>63.</b> (d)	64.	(d)	<b>65.</b> (c)				



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