AspirePrep Education
XI | XII | NEET | IIT-JEE | CUET

NEET 2023

SECTION-A (Q.1 TO Q.35)

1. A bullet is fired from a gun at the speed of 280 ms^{-1} in the direction 30° above the horizontal. The maximum height attained by the bullet is ($g = 9.8\ ms^{-2}$, $\sin 30° = 0.5$)
(a) 2800 m (b) 2000 m (c) 1000 m (d) 3000 m

2. A metal wire has mass (0.4 ± 0.002)g, radius (0.3 ± 0.001)mm and length (5 ± 0.02)cm. The maximum possible percentage error in the measurement of density will nearly be:
(a) 1.2 % (b) 1.3 % (c) 1.6 % (d) 1.4 %

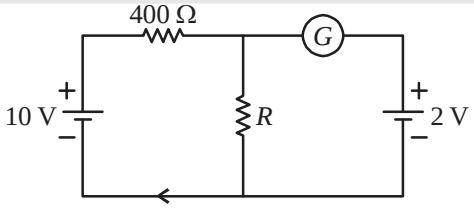
3. Given below are two statements:
Statement-1: Photovoltaic devices can convert optical radiation into electricity.
Statement-2: Zener diode is designed to operate under reverse bias in breakdown region.
In the light of the above statements, choose the most appropriate answer from the options given below:
(a) Statement-1 is incorrect but Statement-2 is correct.
(b) Statement-1 and Statement-2 are incorrect.
(c) Statement-1 is correct but statement-2 is incorrect.
(d) Statement-1 is incorrect but statement-2 is correct.

4. The magnetic energy stored in an inductor of inductance 4 μH carrying a current of 2A is
(a) 4 μJ (b) 4 mJ (c) 8 mJ (d) 8 μJ

5. If $\oint_S \vec{E} \cdot d\vec{S} = 0$ over a surface, then:
(a) the number of flux lines entering the surface must be equal to the number of flux lines leaving it.
(b) the magnitude of electric field on the surface is constant.
(c) all the charges must necessarily be inside the surface.
(d) the electric field inside the surface is necessarily uniform.

6. The net magnetic flux through any closed surface is
(a) Zero (b) Positive (c) Infinity (d) Negative

7. If the galvanometer *G* does not show any deflection in the circuit shown, the value of *R* is given by



(a) 200 Ω (b) 50 Ω (c) 100 Ω (d) 400 Ω

8. In a series LCR circuit, the inductance L is 10 mH, capacitance C is 1 μF and resistance R is 100 Ω . The frequency at which resonance occurs is:
(a) 15.9 rad/s (b) 15.9 kHz (c) 1.59 rad/s (d) 1.59 kHz

9. The temperature of a gas is –50 ºC. To what temperature the gas should be heated so that the r.m.s. speed is increased by 3 times ?
(a) 669 ºC (b) 3295 ºC (c) 3097 K (d) 223 K

10. Let a wire be suspended from the ceiling (rigid support) and stretched by a weight W attached at its free end. The longitudinal stress at any point of cross-sectional area A of the wire is
(a) 2W/A (b) W/A (c) W/2A (d) Zero

11. A Carnot engine has an efficiency of 50 % when its source is at a temperature 327 ºC. The temperature of the sink is
(a) 27 ºC (b) 15 ºC (c) 100 ºC (d) 200 ºC

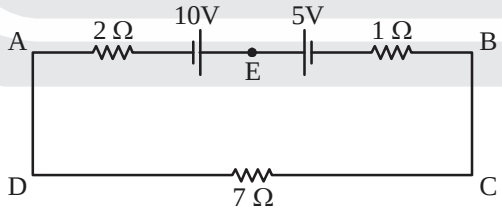
12. Resistance of a carbon resistor determined from colour codes is (22000 ± 5%) Ω. The colour of third band must be:
(a) Red (b) Green (c) Orange (d) Yellow

13. The minimum wavelength of X-rays produced by an electron accelerated through a potential difference of V volts is proportional to
(a) $\sqrt{V}$ (b) $\frac{1}{V}$ (c) $\frac{1}{\sqrt{V}}$ (d) V^2

14. For Young's double slit experiment, two statements are given below:
Statement-1: If screen is moved away from the plane of slits, angular separation of the fringes remains constant.
Statement-2: If the monochromatic source is replaced by another monochromatic source of larger wavelength, the angular separation of fringes decreases.
In the light of the above statements, choose the correct answer from the options given below:
(a) Both Statement-1 and Statement-2 are true.
(b) Both Statement-1 and Statement-2 are false.
(c) Statement-1 is true but Statement-2 is false.
(d) Statement-1 is false but Statement-2 is true.

15. The work functions of Caesium (Cs), Potassium (K) and Sodium (Na) are 2.14 eV, 2.30 eV and 2.75 eV respectively. If incident electromagnetic radiation has an incident energy of 2.20 eV, which of these photosensitive surfaces may emit photoelectrons ?
(a) Cs only (b) Both Na and K (c) K only (d) Na only

16. The magnitude and direction of the current in the following circuit is



(a) 0.2 A from B to A through E (b) 0.5 A from A to B through E
(c) 5/9 A from A to B through E (d) 1.5 A from B to A through E

17. In hydrogen spectrum, the shortest wavelength in the Balmer series is λ. The shortest wavelength in the Bracket series is
(a) 2λ (b) 4λ (c) 9λ (d) 16λ

18. An electric dipole is placed at an angle of 30° with an electric field of intensity 2×10^5 NC^{-1}. It experiences a torque equal to 4 Nm. Calculate the magnitude of charge on the dipole, if the dipole length is 2 cm.
(a) 8 mC (b) 6 mC (c) 4 mC (d) 2 mC

19. The half-life of a radioactive substance is 20 minutes. In how much time, the activity of substance drops to (1/16)th of its initial value ?
(a) 20 minutes (b) 40 minutes (c) 60 minutes (d) 80 minutes

20. The venturi-meter works on
(a) Huygen's principle (b) Bernoulli's principle
(c) The principle of parallel axes (d) The principle of perpendicular axes

21. An ac source is connected to a capacitor C. Due to decrease in its operating frequency
(a) Capacitive reactance decreases (b) Displacement current increases
(c) Displacement current decreases (c) Capacitive reactance remains constant

22. A football player is moving southward and suddenly turns eastward with the same speed to avoid an opponent. The force that acts on the player while turning is
(a) Along eastward (b) Along northward (c) Along north-east (d) Along south-west

23. A full wave rectifier circuit consists of two p-n junction diodes, a centre-tapped transformer, capacitor and a load resistance. Which of these components remove the ac ripple from the rectified output?
(a) A centre-tapped transformer (b) p-n junction diodes
(c) Capacitor (d) Load resistance

24. The potential energy of a long spring when stretched by 2 cm is U. If the spring is stretched by 8 cm, potential energy stored in it will be
(a) 2 U (b) 4 U (c) 8 U (d) 16 U

25. In a plane electromagnetic wave travelling in free space, the electric field component oscillates sinusoidally at a frequency of 2.0×10^{10} Hz and amplitude 48 V m^{-1}. Then the amplitude of oscillating magnetic field is (Speed of light in free space = 3×10^8 m s^{-1})
(a) 1.6×10^{-9} T (b) 1.6×10^{-8} T (c) 1.6×10^{-7} T (d) 1.6×10^{-6} T

26. Light travels a distance x in time t_1 in air and $10x$ in time t_2 in another denser medium. What is the critical angle for this medium?
(a) $\sin^{-1}\left(\frac{t_2}{t_1}\right)$ (b) $\sin^{-1}\left(\frac{10t_2}{t_1}\right)$ (c) $\sin^{-1}\left(\frac{t_1}{10t_2}\right)$ (d) $\sin^{-1}\left(\frac{10t_1}{t_2}\right)$

27. The amount of energy required to form a soap bubble of radius 2 cm from a soap solution is nearly (surface tension of soap solution = 0.03 N m^{-1})
(a) 30.16×10^{-4} J (b) 5.06×10^{-4} J (c) 3.01×10^{-4} J (d) 50.1×10^{-4} J

28. A 12 V, 60 W lamp is connected to the secondary of a step-down transformer, whose primary is connected to ac mains of 220 V. Assuming the transformer to be ideal, what is the current in the primary winding?
(a) 0.27 A (b) 2.7 A (c) 3.7 A (d) 0.37 A

29. The angular acceleration of a body, moving along the circumference of a circle, is
(a) Along the radius, away from centre (b) Along the radius towards the centre
(c) Along the tangent to its position (d) Along the axis of rotation

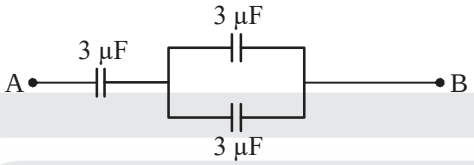
30. A vehicle travels half the distance with speed v and the remaining distance with speed $2v$. Its average speed is
(a) $v/3$ (b) $2v/3$ (c) $4v/3$ (d) $3v/4$

31. The errors in the measurement which arise due to unpredictable fluctuations in temperature and voltage supply are
(a) Instrumental errors (b) Personal errors (c) Least count errors (d) Random errors

32. The ratio of radius of gyration of a solid sphere of mass M and radius R about its own axis to the radius of gyration of the thin hollow sphere of same mass and radius about its axis is
(a) 3 : 5 (b) 5 : 3 (c) 2 : 5 (d) 5 : 2

33. Two bodies of mass m and $9m$ are placed at a distance R. The gravitational potential on the line joining the bodies where the gravitational field equals zero, will be (G = gravitational constant)
(a) $-\frac{8Gm}{R}$ (b) $-\frac{12Gm}{R}$ (c) $-\frac{16Gm}{R}$ (d) $-\frac{20Gm}{R}$

34. The equivalent capacitance of the system shown in the following circuit is



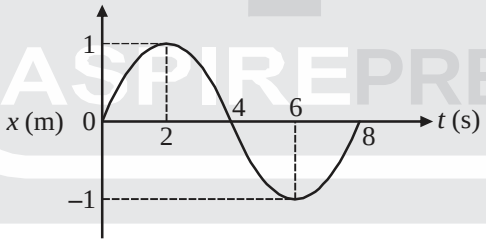
(a) 2 μF (b) 3 μF (c) 6 μF (d) 9 μF

35. The ratio of frequencies of fundamental harmonic produced by an open pipe to that of closed pipe having the same length is
(a) 1 : 2 (b) 2 : 1 (c) 1 : 3 (d) 3 : 1

SECTION-B (Q.36 TO Q.50)

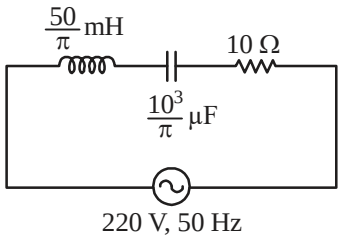
36. Calculate the maximum acceleration of a moving car so that a body lying on the floor of the car remains stationary. The coefficient of static friction between the body and the floor is 0.15 ($g = 10\ m\ s^{-2}$).
(a) $1.2\ m\ s^{-2}$ (b) $150\ m\ s^{-2}$ (c) $1.5\ m\ s^{-2}$ (d) $50\ m\ s^{-2}$

37. The x-t graph of a particle performing simple harmonic motion is shown in the figure.



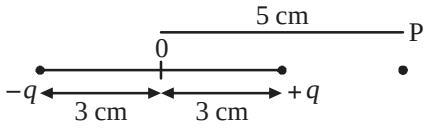
The acceleration of the particle at $t = 2$ s is
(a) $\frac{\pi^2}{8}\ m\ s^{-2}$ (b) $-\frac{\pi^2}{8}\ m\ s^{-2}$ (c) $\frac{\pi^2}{16}\ m\ s^{-2}$ (d) $-\frac{\pi^2}{16}\ m\ s^{-2}$

38. The net impedance of circuit (as shown in figure) will be



(a) $10\sqrt{2}\ \Omega$ (b) $15\ \Omega$ (c) $5\sqrt{5}\ \Omega$ (d) $25\ \Omega$

39. An electric dipole is placed as shown in the figure.



The electric potential (in 10^2 V) at point P due to the dipole is (ε_0 = permittivity of free space and $\frac{1}{4\pi\varepsilon_0} = K$)

(a) $\left(\frac{3}{8}\right)qK$ (b) $\left(\frac{5}{8}\right)qK$ (c) $\left(\frac{8}{5}\right)qK$ (d) $\left(\frac{8}{3}\right)qK$

40. A wire carrying a current I along the positive x-axis has length L. It is kept in a magnetic field

$$\vec{B} = (2\hat{i} + 3\hat{j} - 4\hat{k})\ \text{T}$$

The magnitude of the magnetic force acting on the wire is

(a) $3IL$ (b) $\sqrt{5}\,IL$ (c) $5IL$ (d) $\sqrt{3}\,IL$

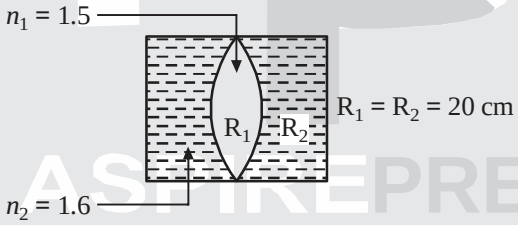
41. A bullet from a gun is fired on a rectangular wooden block with velocity u. When bullet travels 24 cm through the block along its length horizontally, velocity of bullet becomes $u/3$. Then it further penetrates into the block in the same direction before coming to rest exactly at the other end of the block. The total length of the block is

(a) 27 cm (b) 24 cm (c) 28 cm (d) 30 cm

42. Two thin lenses are of same focal lengths (f), but one is convex and the other one is concave. When they are placed in contact with each other, the equivalent focal length of the combination will be

(a) Zero (b) $f/4$ (c) $f/2$ (d) Infinite

43. In the figure shown here, what is the equivalent focal length of the combination of lenses (Assume that all layers are thin)?

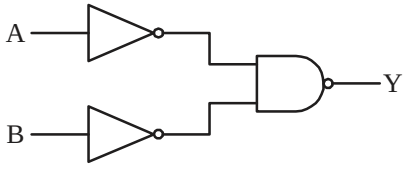


(a) 40 cm (b) – 40 cm (c) –100 cm (d) –50 cm

44. A satellite is orbiting just above the surface of the earth with period T. If d is the density of the earth and G is the universal constant of gravitation, the quantity $3\pi/Gd$ represents:

(a) T (b) T^2 (c) T^3 (d) $\sqrt{T}$

45. For the following logic circuit, the truth table is



(a)

A	B	Y
0	0	1
0	1	1
1	0	1
1	1	0

(b)

A	B	Y
0	0	0
0	1	1
1	0	1
1	1	1

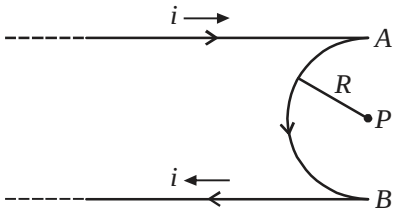
(c)

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	0

(d)

A	B	Y
0	0	0
0	1	0
1	0	0
1	1	1

46. A very long conducting wire is bent in a semi-circular shape from *A* to *B* as shown in figure.



The magnetic field at point *P* for steady current configuration is given by

(a) $\frac{\mu_0 i}{4R}$ pointed into the page (b) $\frac{\mu_0 i}{4R}$ pointed away from the page

(c) $\frac{\mu_0 i}{4R}\left[1-\frac{2}{\pi}\right]$ pointed away from the page (d) $\frac{\mu_0 i}{4R}\left[1-\frac{2}{\pi}\right]$ pointed into the page

47. The resistance of platinum wire at 0ºC is 2Ω and 6.8Ω at 80ºC. The temperature coefficient of resistance of the wire is

(a) 3×10^{-4} ºC^{-1} (b) 3×10^{-3} ºC^{-1} (c) 3×10^{-2} ºC^{-1} (d) 3×10^{-1} ºC^{-1}

48. 10 resistors, each of resistance *R* are connected in series to a battery of emf *E* and negligible internal resistance. Then those are connected in parallel to the same battery, the current is increased n times. The value of *n* is

(a) 10 (b) 100 (c) 1 (d) 1000

49. The radius of inner most orbit of hydrogen atom is 5.3×10^{-11} m. What is the radius of third allowed orbit of hydrogen atom?

(a) 0.53 Å (b) 1.06 Å (c) 1.59 Å (d) 4.77 Å

50. A horizontal bridge is built across a river. A student standing on the bridge throws a small ball vertically upwards with a velocity 4 m s^{-1}. The ball strikes the water surface after 4 s. The height of bridge above water surface is (Take g = 10 m s^{-2})

(a) 56 m (b) 60 m (c) 64 m (d) 68 m

SECTION-B (Q.51 TO Q.100)

51. The relation between n_m, (n_m = the number of permissible values of magnetic quantum number(m)) for a given value of azimuthal quantun number (*l*), is

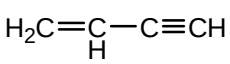
(a) $\ell = \frac{n_m - 1}{2}$ (b) $\ell = 2n_m + 1$ (c) $n_m = 2\ell^2 + 1$ (d) $n_m = \ell + 2$

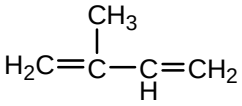
52. The element expected to form largest ion to achieve the nearest noble gas configuration is

(a) O (b) F (c) N (d) Na

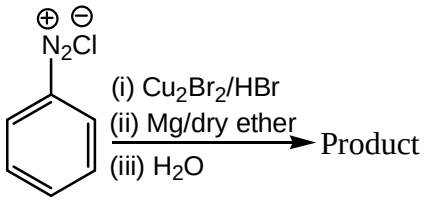
53. Which amongst the following molecules on polymerization produces neoprene?

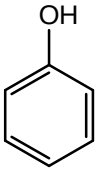
(a) $H_2C{=}\underset{H}{C}{-}\underset{H}{C}{=}CH_2$ (b) $H_2C{=}\overset{Cl}{\overset{|}{C}}{-}\underset{H}{C}{=}CH_2$

(c) 

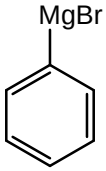
(d) 

54. Identify the product in the following reaction:



(a) 

(b) 

(c) 

(d) 

55. The conductivity of centimolar solution of KCl at 25°C is 0.0210 $ohm^{-1}cm^{-1}$ and the resistance of the cell containing the solution at 5°C is 60 ohm. The value of cell constant is

(a) 1.34 cm^{-1} (b) 3.28 cm^{-1} (c) 1.26 cm^{-1} (d) 3.34 cm^{-1}

56. Match **List-I** with **List-II**:

(A) Coke	(I) Carbon atoms are sp^3 hybridised
(B) Diamond	(II) Used as a dry lubricant
(C) Fullerene	(III) Used as a reducing agent
(D) Graphite	(IV) Cage like molecules

Choose the **correct** answer from the options given below:

(a) A-II, B-IV, C-I, D-III (b) A-IV, B-I, C-II, D-III

(c) A-III, B-I, C-IV, D-II (d) A-III, B-IV, C-I, D-II

57. Given below are two statements one is labelled as **Assertion A** and the other is labelled as **Reason R**:

Assertion A: A reaction can have zero activation energy.

Reason R: The minimum extra amount of energy absorbed by reactant molecules so that their energy becomes equal to threshold value, is called activation energy

In the light of the above statements, choose the **correct** answer from the options given below:

(a) Both **A** and **R** are true and **R** is the correct explanation of **A**

(b) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**

(c) **A** is true but **R** is false

(d) **A** is false but **R** is true.

58. Homoleptic complex from the following complexes is

(a) Potassium trioxalatoaluminate (III) (b) Diamminechloridonitrito -N-platinum (II)

(c) Pentaamminecarbonatocobalt (III) chloride (d) Triamminetriaquachromium (III) chloride

59. A compound is formed by two elements A and B. The element B forms cubic close packed structure and atoms of A occupy 1/3 of tetrahedral voids. If the formula of the compound is A_xB_y, then the value of x + y is in option

(a) 5 (b) 4 (c) 3 (d) 2

60. The right option for the mass of CO_2 produced by heating 20g of 20% pure limestone is (Atomic mass of Ca = 40)

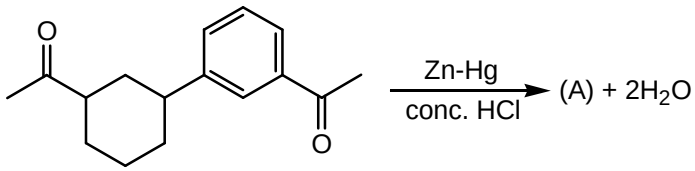
$$\left[CaCO_3 \xrightarrow{1200\ K} CaO + CO_2 \right]$$

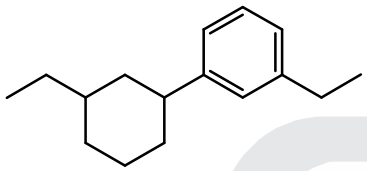
(a) 1.12 g (b) 1.76 g (c) 2.64 g (d) 1.32 g

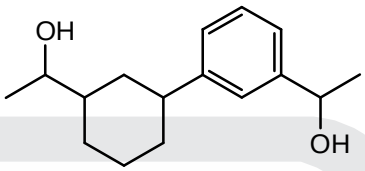
61. Taking stability as the factor, which one of the following represents correct relationship?

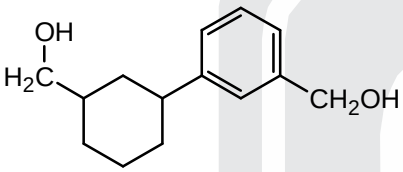
(a) $TlCl_3 > TlCl$ (b) $InI_3 > InI$ (c) $AlCl > AlCl_3$ (d) $TlI > TlI_3$

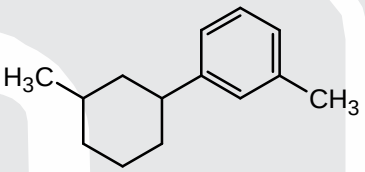
62. Identify product (A) in the following reaction:



(a) 

(b) 

(c) 

(d) 

63. Select the **correct** statements from the following:

(A) Atoms of all elements are composed of two fundamental particles

(B) The mass of the electron is 9.10939×10^{-31} kg

(C) All the isotopes of a given element show same chemical properties

(D) Protons and electrons are collectively known as ncucleons

(E) Dalton's atomic theory, regarded the atom as an ultimate particle of matter.

Choose the **correct** answer from the options given below:

(a) A, B and C only (b) C, D and E only (c) A and E only (d) B, C and E only

64. For a certain reaction, the rate = $k[A]^2[B]$, when the initial concentration of A is tripled keeping concentration of B constant, the initial rate would

(a) decrease by a factor of nine (b) increase by a factor of six

(c) increase by a factor of nine (d) increase by a factor of three

65. Given below are two statements:

Statement-I: A unit formed by the attachment of a base to $1'$ position of sugar is known as nucleoside

Statement-II: When nucleoside is linked to phosphorous acid at $5'$-position of sugar moiety, we get nucleotide

In the light of the above statements, choose the correct answer from the options given below.

(a) Both Statement I and Statement II are true (b) Both Statement I and Statement II are false

(c) Statement I is true Statement II is false (d) Statement I is false but Statement is true

66. Amongst the given options which of the following molecules/ion acts as a Lewis acid?

(a) NH_3 (b) H_2O (c) BF_3 (d) OH^-

67. Which of the following reactions will NOT give primary amine as the product?

(a) $CH_3CONH_2 \xrightarrow{Br_2/KOH}$ Product (b) $CH_3CN \xrightarrow[(ii)\ H_3O^+]{(i)\ LiAlH_4}$ Product

(c) $CH_3NC \xrightarrow[(ii)\ H_3O]{(i)\ LiAlH_4}$ Product (d) $CH_3CONH_2 \xrightarrow[(ii)\ H_3O^+]{(i)\ LiAlH_4}$ Product

68. Amongst the following, the total number of species NOT having eight electrons around central atom in its outer most shell, is NH_3, $AlCl_3$, $BeCl_2$, CCl_4, PCl_5.

(a) 3 (b) 2 (c) 4 (d) 1

69. Which of the following statements are NOT correct?

(A) Hydrogen is used to reduce heavy metal oxides to metals

(B) Heavy water is used to study reaction mechanism

(C) Hydrogen is used to make saturated fats from oils

(D) The H-H bond dissociation enthalpy is lowest as compared to a single bond between two atoms of any element.

(E) Hydrogen reduces oxides of metals that are more active than iron.

Choose the **most appropriate** answer from the options given below:

(a) B, C, D, E only (b) B, D only

(c) D, E only (d) A, B, C only

70. Weight (g) of two moles of the organic compound, which is obtained by heating sodium ethanoate with sodium hydroxide in presenec of calcium oxide is

(a) 16 (b) 32 (c) 30 (d) 18

71. The number of σ bonds, π bonds and lone pair of electrons in pyridine, respectively are:

(a) 11, 2, 0 (b) 12, 3, 0 (c) 11, 3, 1 (d) 12, 2, 1

72. Which one of the following statements is **correct**?

(a) The daily requirement of Mg and Ca in the human body is estimated to be 0.2 – 0.3 g.

(b) All enzymes that utilise ATP in phosphate transfer require Ca as the cofactor

(c) The bone in human body is an inert and unchanging substance.

(d) Mg plays roles in neuromuscular function and interneuronal transmission

73. Given below are two statements one is labelled as **Assertion A** and the other is labelled as **Reason R**:

Assertion A: Metallic sodium dissolves in liquid ammonia giving a deep blue solution, which is paramgnetic

Reason R: The deep blue solution is due to the formation of amide.

In the light of the above statements, choose the **correct** answer from the options given below:

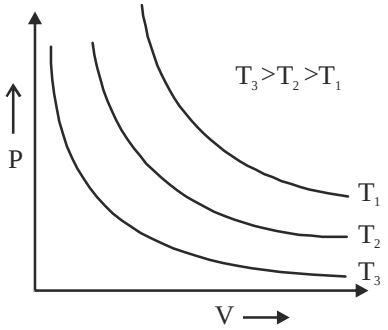
(a) Both **A** and **R** are true and **R** is the correct explanation of **A**

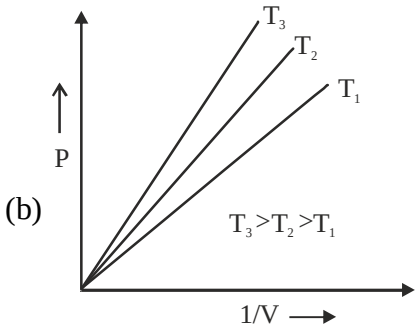
(b) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**

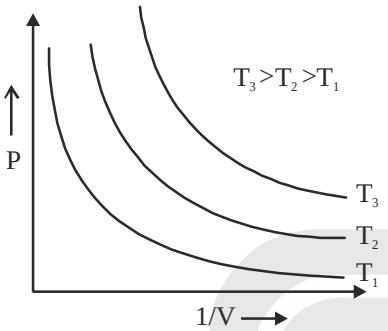
(c) **A** is true but **R** is false

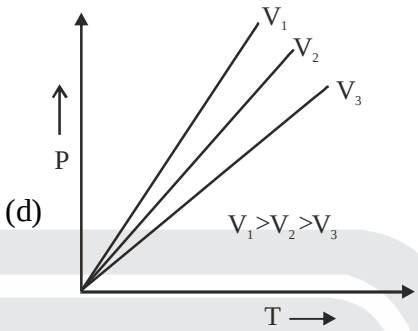
(d) **A** is false but **R** is true.

74. Which amongst the following options is **correct** graphical representation of Boyle's Law?

(a) 

(b) 

(c) 

(d) 

75. The correct order of energies of molecules orbitals of N_2 molecule, is

(a) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \left(\pi 2p_x = \pi 2p_y\right) < \sigma 2p_z < \left(\pi^* 2p_x = \pi^* 2p_y\right) < \sigma^* 2p_z$

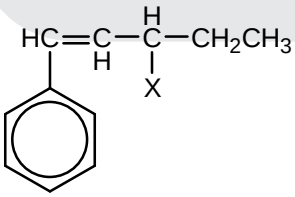
(b) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \sigma 2p_z < \left(\pi 2p_x = \pi 2p_y\right) < \left(\pi^* 2p_x = \pi^* 2p_y\right) < \sigma^* 2p_z$

(c) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \sigma 2p_z < \sigma^* 2p_z < \left(\pi 2p_x = \pi 2p_y\right) < \left(\pi^* 2p_x = \pi^* 2p_y\right)$

(d) $\sigma 1s < \sigma^* 1s < \sigma 2s < \sigma^* 2s < \left(\pi 2p_x = \pi 2p_y\right) < \left(\pi^* 2p_x = \pi^* 2p_y\right) < \sigma 2p_z < \sigma^* 2p_z$

76. Some tranquilizers are listed below. Which one from the following belongs to barbiturates?
(a) Chlordiazepoxide (b) Meprobamate (c) Valium (d) Veronal

77. The given compound,



is an example of _____
(a) benzylic halide (b) aryl halide
(c) allylic halide (d) vinylic halide

78. Intermolecular forces are forces of attraction and repulsion between interacting particles that will include:
(A) dipole-dipole forces. (B) dipole-induced dipole forces.
(C) hydrogen bonding. (D) covalent bonding.
(E) dispersion forces.

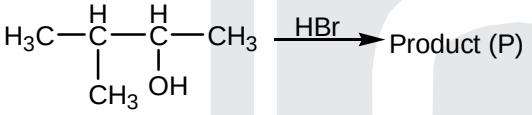
Choose the most appropriate answer from the options given below:
(a) B, C, D, E are correct. (b) A, B, C, D are correct.
(c) A. B, C, E are correct. (d) A, C, D, E are correct.

79. Which one is an example of heterogenous catalysis?
(a) Oxidation of sulphur dioxide into sulphur trioxide in the presence of oxides of nitrogen.
(b) Hydrolysis of sugar catalysed by H^+ ions.
(c) Decomposition of ozone in presence of nitrogen monoxide.
(d) Combination between dinitrogen and dihydrogen to form ammonia in the presence of finely divided iron.

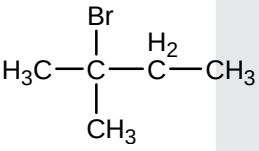
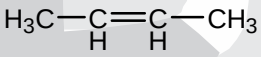
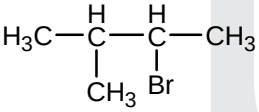
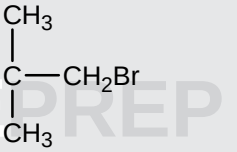
80. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R:**
Assertion A: Helium is used to dilute oxygen in diving apparatus.
Reasons R: Helium has high solubility in O_2.
In the light of the above statements, choose the correct answer from the options given below:
(a) Both **A** and **R** are true and **R** is the correct explanation of **A**.
(b) Both **A** and **R** are true and **R** is NOT the correct explanation of **A**.
(c) **A** is true but **R** is false.
(d) **A** is false but **R** is true.

81. The stability of Cu^{2+} is more than Cu^+ salts in aqueous solution due to
(a) first ionisation enthalpy. (b) enthalpy of atomization.
(c) hydration energy. (d) second ionisation enthalpy.

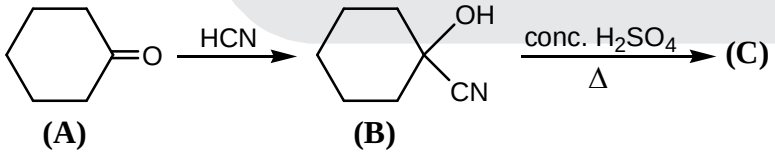
82. Consider the following reaction and identify the product (P)



3-Methylbutan-2-ol

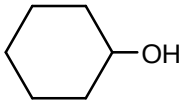
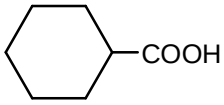
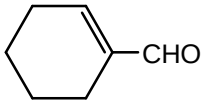
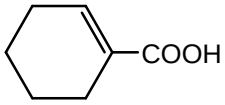
(a) 
(b) 
(c) 
(d) 

83. Compelete the following reaction:



(A) **(B)**

(C) is _____

(a)  (b)  (c)  (d) 

84. Given below are two statements one is labelled as **Assertion A** and the other is labelled as **Reason R**:
Assertion A: In equation $\Delta_r G = -nFE_{cell}$, value of $\Delta_r G$ depends on n.
Reason R: E_{cell} is is an intensive property and $\Delta_r G$ is an extensive property.

In the light of the above statements, choose the **correct** answer from the options given below:

(a) Both **A** and **R** are true and **R** is the correct explanation of **A**

(b) Both **A** and **R** are true and **R** is **NOT** the correct explanation of **A**

(c) **A** is true but **R** is false

(d) **A** is false but **R** is true.

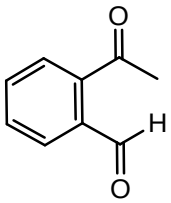
85. In Lassaigne's extract of an organic compound, both nitrogen and sulphur are present, which gives blood red colour with Fe^{3+} due to the formation of

(a) $Fe_4\left[Fe(CN)_6\right]_3.xH_2O$ (b) NaSCN

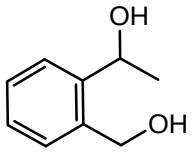
(c) $\left[Fe(CN)_5 NOS\right]^{4-}$ (d) $\left[Fe(SCN)\right]^{2+}$

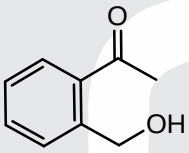
Chemistry: Section-B (Q.No. 86 to 100)

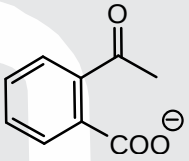
86. Identify the major product obtained in the following reaction:

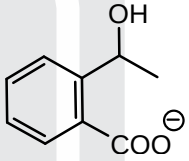


+ $2[Ag(NH_3)_2]^+ + 3\ OH^- \xrightarrow{\Delta}$ major product

(a) 

(b) 

(c) 

(d) 

87. Match **List-I** with **List-II**

List-I (Oxoacids of Sulphur)	**List-II (Bonds)**
(A) Peroxodisulphuric acid	(I) Two S-OH, Four S=O, One S-O-S
(B) Sulphuric acid	(II) TWo S-OH, One S=O
(C) Pyrosulphuric acid	(III) Two S-OH, Four S=O, One S-O-O-S
(D) Sulphurous acid	(IV) Two S-OH, Two S=O

Choose the **correct** answer from the options given below:

(a) A-I, B-III, C-II, D-IV (b) A-III, B-IV, C-I, D-II

(c) A-I, B-III, C-IV, D-II (d) A-III, B-IV, C-II, D-I

88. On balancing the given redox reaction,

$$aCr_2O_7^{2-} + bSO_3^{2-}(aq) + cH^+(aq) \rightarrow 2aCr^{3+}(aq) + bSO_4^{2-}(aq) + \frac{c}{2}H_2O(\ell)$$

the coefficients a, b and c are found to be, respectively

(a) 1, 3, 8 (b) 3, 8, 1 (c) 1, 8, 3 (d) 8, 1, 3

89. Pumice stone is an example of

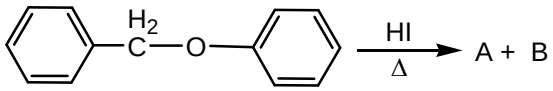
(a) sol (b) gel (c) solid sol (d) foam

90. Which complex compound is most stable?

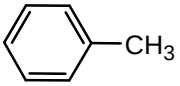
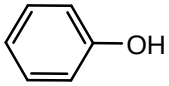
(a) $\left[Co(NH_3)_4(H_2O)Br\right](NO_3)_2$ (b) $\left[Co(NH_3)_3(NO_3)_3\right]$

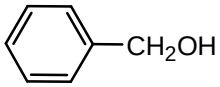
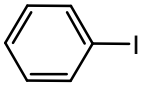
(c) $\left[CoCl_2(en)_2\right]NO_3$ (d) $\left[Co(NH_3)_6\right]_2(SO_4)_3$

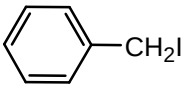
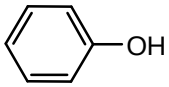
91. Consider the following reaction:

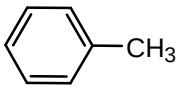
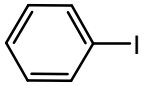


Identify products A and B

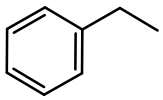
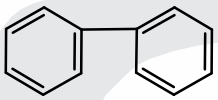
(a) A =  and B = 

(b) A =  and B = 

(c) A =  and B = 

(d) A =  and B = 

92. Identify the final product [D] obtained in the following sequence of reactions

(a)  (b)  (c) C_4H_{10} (d)

93. Given below are two statements:

Statement-I: The nutrient deficient water bodies lead to eutrophication

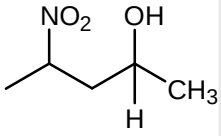
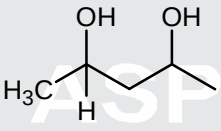
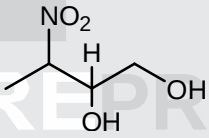
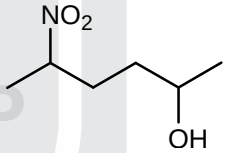
Statement-II: Eutrophication leads to decrease in the level of oxygen in the water bodies.

In the light of the above statements, choose the correct answer from the options given below.

(a) Both Statement I and Statement II are true (b) Both Statement I and Statement II are false

(c) Statement I is true Statement II is false (d) Statement I is false but Statement is true

94. Which amongst the following will be most readily dehydrated under acidic conditions?

(a)  (b)  (c)  (d) 

95. What fraction of one edge centred octahedral void lies in one unit cell of Fcc?

(a) 1/2 (b) 1/3 (c) 1/4 (d) 1/12

96. The reaction that does **NOT** take place in a blast furnace between 900 K to 1500 K temperature range during extraction of iron is

(a) $Fe_2O_3 + CO \rightarrow 2FeO + CO_2$ (b) $FeO + CO \rightarrow Fe + CO_2$

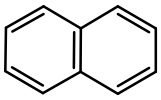
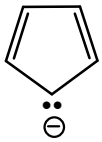
(c) $C + CO_2 \rightarrow 2CO$ (d) $CaO + SiO_2 \rightarrow CaSiO_3$

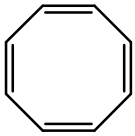
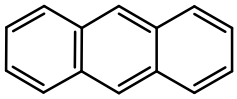
97. Which amongst the following options is the correct relation between change in enthalpy and change in internal energy?

(a) $\Delta H = \Delta U - \Delta n_g RT$ (b) $\Delta H = \Delta U + \Delta n_g RT$

(c) $\Delta H - \Delta U = -\Delta n_g RT$ (d) $\Delta H + \Delta U = \Delta n_g RT$

98. Consider the following compounds/species:

(I)  (II)  (III)  (IV) 

(V)  (VI)  (VII) 

The number of compounds/species which obey Huckel's rule is

(a) 4 (b) 6 (c) 2 (d) 5

99. The equilibrium concentrations of the species in the reaction $A + B \rightleftharpoons C + D$ are 2, 3, 10 and 6 mol L^{-1}, respectively at 300K. ΔG^0 for the reaction is (R = 2 cal/mol K)

(a) 1372.60 cal (b) –137.26 cal (c) –1381.80 cal (d) –13.73 cal

100. Which of the following statements are **INCORRECT**?

(A) All the transition metals except scandium form MO oxides which are ionic

(B) The highest oxidation number corresponding to the group number in transition metal oxides is attained in Sc_2O_3 to Mn_2O_7.

(C) Basic character increases from V_2O_3 to V_2O_4 to V_2O_5.

(D) V_2O_4 dissolves in acids to give VO_4^{3-} salts.

(E) CrO is basic but Cr_2O_3 is amphoteric.

Choose the **correct** answer from the options given below:

(a) A and E only (b) B and D only (c) C and D only (d) B and C only

101. Movement and accumulation of ions across a membrane against their concentration gradient can be explained by

(a) Osmosis (b) Facilitated Diffusion (c) Passive Transport (d) Active Transport

102. How many ATP and NADPH2 are required for the synthesis of one molecule of Glucose during Calvin cycle?

(a) 12 ATP and 12 $NADPH_2$ (b) 18 ATP and 12 $NADPH_2$

(c) 12 ATP and 16 $NADPH_2$ (d) 18 ATP and 16 $NADPH_2$

103. Which micronutrient is required for splitting of water molecule during photosynthesis?

(a) Manganese (2) Molybdenum (3) Magnesium (4) Copper

104. Given below are two statements :

Statement I: Endarch and exarch are the terms often used for describing the position of secondary xylem in the plant body.

Statement II: Exarch condition is the most common feature of the root system.

In the light of the above statements, choose the correct answer from the options given below:

(1) Both Statement I and Statement II are true (2) Both Statement I and Statement II are false

(3) Statement I is correct but Statement II is false (4) Statement I is incorrect but Statement II is true

105. Given below are two statements : One is labelled as Assertion A and the other is labelled as Reason R :

Assertion A : Late wood has fewer xylary elements with narrow vessels.

Reason R: Cambium is less active in winters.

In the light of the above statements, choose the correct answer from the options given below :

(1) Both A and R are true and R is the correct explanation of A
(2) Both A and R are true but R is NOT the correct explanation of A
(3) A is true but R is false
(4) A is false but R is true

106. During the purification process for recombinant DNA technology, addition of chilled ethanol precipitates out
(1) RNA (2) DNA (3) Histones (4) Polysaccharides

107. In gene gun method used to introduce alien DNA into host cells, microparticles of _____ metal are used.
(1) Copper (2) Zinc
(3) Tungsten or gold (4) Silver

108. The historic Convention on Biological Diversity, 'The Earth Summit' was held in Rio de Janeiro in the year
(1) 1985 (2) 1992 (3) 1986 (4) 2002

109. In angiosperm, the haploid, diploid and triploid structures of a fertilized embryo sac sequentially are :
(1) Synergids, Primary endosperm nucleus and zygote
(2) Antipodals, synergids, and primary endosperm nucleus
(3) Synergids, Zygote and Primary endosperm nucleus
(4) Synergids, antipodals and Polar nuclei

110. Upon exposure to UV radiation, DNA stained with ethidium bromide will show
(1) Bright red colour (2) Bright blue colour
(3) Bright yellow colour (4) Bright orange colour

111. Family Fabaceae differs from Solanaceae and Liliaceae. With respect to the stamens, pick out the characteristics specific to family Fabaceae but not found in Solanaceae or Liliaceae.
(1) Diadelphous and Dithecous anthers
(2) Polyadelphous and epipetalous stamens
(3) Monoadelphous and Monothecous anthers
(4) Epiphyllous and Dithecous anthers

112. Which hormone promotes internode/petiole elongation in deep water rice?
(1) GA_3 (2) Kinetin (3) Ethylene (4) 2, 4-D

113. Frequency of recombination between gene pairs on same chromosome as a measure of the distance between genes to map their position on chromosome, was used for the first time by
(1) Thomas Hunt Morgan (2) Sutton and Boveri
(3) Alfred Sturtevant (4) Henking

114. Identify the pair of heterosporous pteridophytes among the following :
(1) Lycopodium and Selaginella (2) Selaginella and Salvinia
(3) Psilotum and Salvinia (4) Equisetum and Salvinia

115. What is the role of RNA polymerase III in the process of transcription in Eukaryotes?
(1) Transcription of rRNAs (28S, 18S and 5.8S)
(2) Transcription of tRNA, 5S rRNA and snRNA
(3) Transcription of precursor of mRNA
(4) Transcription of only snRNAs

116. Expressed Sequence Tags (ESTs) refers to
(1) All genes that are expressed as RNA. (2) All genes that are expressed as proteins.
(3) All genes whether expressed or unexpressed. (4) Certain important expressed genes.

117. Cellulose does not form blue colour with Iodine because
(1) It is a disaccharide
(2) It is a helical molecule
(3) It does not contain complex helices and hence cannot hold iodine molecules
(4) It breaks down when iodine reacts with it

118. Given below are two statements : One is labelled as Assertion A and the other is labelled as Reason R :
Assertion A: ATP is used at two steps in glycolysis.
Reason R: First ATP is used in converting glucose into glucose-6-phosphate and second ATP is used in conversion of fructose-6-phosphate into fructose-1, 6-diphosphate.
In the light of the above statements, choose the correct answer from the options given below :
(1) Both A and R are true and R is the correct explanation of A.
(2) Both A and R are true but R is NOT the correct explanation of A.
(3) A is true but R is false.
(4) A is false but R is true

119. The phenomenon of pleiotropism refers to
(1) Presence of several alleles of a single gene controlling a single crossover
(2) Presence of two alleles, each of the two genes controlling a single trait
(3) A single gene affecting multiple phenotypic expression
(4) More than two genes affecting a single character

120. Identify the correct statements:
A. Detrivores perform fragmentation.
B. The humus is further degraded by some microbes during mineralization.
C. Water soluble inorganic nutrients go down into the soil and get precipitated by a process called leaching.
D. The detritus food chain begins with living organisms.
E. Earthworms break down detritus into smaller particles by a process called catabolism.
Choose the correct answer from the options given below:
(1) A, B, C only (2) B, C, D only (3) C, D, E only (4) D, E, A only

121. The thickness of ozone in a column of air in the atmosphere is measured in terms of :
(1) Dobson units (2) Decibels (3) Decameter (4) Kilobase

122. Given below are two statements :
Statement I: The forces generated transpiration can lift a xylem-sized column of water over 130 meters height.
Statement II: Transpiration cools leaf surfaces sometimes 10 to 15 degrees evaporative cooling.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Both Statement I and Statement II are correct
(2) Both Statement I and Statement II are incorrect
(3) Statement I is correct but Statement II is incorrect
(4) Statement I is incorrect but Statement II is correct

123. What is the function of tassels in the corn cob?
(1) To attract insects (2) To trap pollen grains
(3) To disperse pollen grains (4) To protect seeds

124. The reaction centre in PS II has an absorption maxima at
(1) 680 nm (2) 700 nm (3) 660 nm (4) 780 nm

125. In tissue culture experiments, leaf mesophyll cells are put in a culture medium to form callus. This phenomenon may be called as

(1) Differentiation (2) Dedifferentiation (3) Development (4) Senescence

126. Given below are two statements : One labelled as Assertion A and the other labelled as Reason R:

Assertion A: The first stage of gametophyte in the life cycle of moss is protonema stage.

Reason R: Protonema develops directly from spores produced in capsule.

In the light of the above statements, choose the most appropriate answer from options given below:

(1) Both A and R are correct and R is the correct explanation of A

(2) Both A and R are correct but R is NOT the correct explanation of A

(3) A is correct but R is not correct

(4) A is not correct but R is correct

127. Among eukaryotes, replication of DNA takes place in :

(1) M phase (2) S phase (3) G_1 phase (4) G_2 phase

Q. 128 - To Q. 150 (Questions Missing)

151. Match List I with List II.

List I	**List II**
A. Vasectomy	I. Oral method
B. Coitus interruptus	II. Barrier method
C. Cervical caps	III. Surgical method
D. Saheli	IV. Natural method

Choose the ***correct*** answer from the options given below:

(a) A-III, B-I, C-IV, D-II (b) A-III, B-IV, C-I, D-I

(c) A-II, B-III, C-I, D-IV (d) A-IV, B-II, C-I, D-II

152. Given below are two statements:

Statement I: Vas deferens receives a duct from seminal vesicle and opens into urethra as the ejaculatory duct.

Statement II: The cavity of the cervix is called cervical canal which along with vagina forms birth canal.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are true.

(b) Both **Statement I** and **Statement II** are false

(c) **Statement I** is correct but **Statement II** is false

(d) **Statement I** incorrect but **Statement II** is true

153. Which of the following statements is ***correct***?

(a) Eutrophication refers to increase in domestic sewage and waste water in lakes

(b) Biomagnification refers to increase in concentration of the toxicant at successive trophic levels

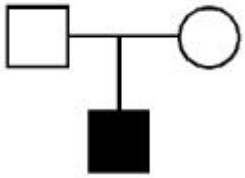
(c) Presence of large amount of nutrients in water restricts 'Algal Bloom'

(d) Algal Bloom decreases fish mortality

154. Which one of the following symbols represents mating between relatives in human pedigree analysis?

(a) 

(b) 

(c) 

(d) 

155. Which one of the following common sexually transmitted diseases is completely curable when detected early and treated properly?

(a) Genital herpes
(b) Gonorrhoea
(c) Hepatitis-B
(d) HIV Infection

156. Match List I with List II.

List I	**List II**
A. Heroin	I. Effect on cardiovascular system
B. Marijuana	II. Slow down body function
C. Cocaine	III. Painkiller
D. Morphine	IV. Interfere with transport of dopamine

Choose the ***correct*** answer from the options given below:

(a) A-II, B-I, C-IV, D-III
(b) A-I, B-II, C-III, D-IV
(c) A-IV, B-III, C-II, D-I
(d) A-III, B-IV, C-I, D-II

157. Match List I with List II

List I (Type of Joint)	**List II (Found between)**
A. Cartilaginous Joint	I. Between flat skull bones
B. Ball and Socket Joint	II. Between adjacent vertebrae in vertebral column
C. Fibrous Joint	III. Between carpal and metacarpal of thumb
D. Saddle Joint	IV. Between Humerus and Pectoral girdle

Choose the ***correct*** answer from the options given below:

(a) A-III, B-I, C-II, D-IV
(b) A-II, B-IV, C-I, D-III
(c) A-I, B-IV, C-III, D-II
(d) A-II, B-IV, C-III, D-I

158. Given below are two statements:

Statement I: A protein is imagined as a line, the left end represented by first amino acid (C-terminal) and the right end represented by last amino acid (N-terminal).

Statement II: Adult human haemoglobin, consists of 4 subunits (two subunits of α type and two subunits of β type)

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are true.
(b) Both **Statement I** and **Statement II** are false.
(c) **Statement I** is true but **Statement II** is false.
(d) **Statement I** is false but **Statement II** is true.

159. Which of the following are NOT considered as the part of endomembrane system?

A. Mitochondria
B. Endoplasmic Reticulum
C. Chloroplasts
D. Golgi complex
E. Peroxisomes

Choose the **most appropriate** answer from the options given below:

(a) B and D only
(b) A, C and E only
(c) A and D only
(d) A, D and E only

160. Given below are two statements:

Statement I: RNA mutates at a faster rate.

Statement II: Viruses having RNA genome and shorter life span mutate and evolve faster.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are true.
(b) Both **Statement I** and **Statement II** are false.
(c) **Statement I** is true but **Statement II** is false.
(d) **Statement I** false but **Statement II** is true.

161. Match List I with List II.

List I	**List II**
A. CCK	I. Kidney
B. GIP	II. Heart
C. ANF	III. Gastric gland
D. ADH	IV. Pancreas

Choose the ***correct*** answer from the options given below:

(a) A-IV, B-III, C-II, D-I
(b) A-III, B-II, C-IV, D-I
(c) A-II, B-IV, C-I, D-III
(d) A-IV, B-II, C-III, D-I

162. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A: Endometrium is necessary for implantation of blastocyst.

Reason R: In the absence of fertilization, the corpus luteum degenerates that causes disintegration of endometrium.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **A** and **R** are true and **R** is the correct explanation of **A**.
(b) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.
(c) **A** is true but **R** is false.
(d) **A** is false but **R** is true.

163. Match List I with List II.

List I	**List II**
A. Ringworm	I. *Haemophilus influenzae*
B. Filariasis	II. *Trichophyton*
C. Malaria	III. *Wuchereria bancrofti*
D. Pneumonia	IV. *Plasmodium vivax*

Choose the ***correct*** answer from the options given below:

(a) A-II, B-III, C-IV, D-I
(b) A-II, B-III, C-I, D-IV
(c) A-III, B-II, C-I, D-IV
(d) A-III, B-II, C-IV, D-I

164. Given below are two statements:

Statement I: Low temperature preserves the enzyme in a temporarily inactive state whereas high temperature destroys enzymatic activity because proteins are denatured by heat.

Statement II: When the inhibitor closely resembles the substrate in its molecular structure and inhibits the activity of the enzyme, it is known as competitive inhibitor.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are true.
(b) Both **Statement I** and **Statement II** are false.
(c) **Statement I** is true but **Statement II** is false.
(d) **Statement I** is false but **Statement II** is true.

165. Match List I with List II.

List I	**List II**
A. *Taenia*	I. Nephridia
B. *Paramoecium*	II. Contractile vacuole
C. *Periplaneta*	III. Flame cells
D. *Pheretima*	IV. Urecose gland

Choose the ***correct*** answer from the options give below:

(a) A-I, B-II, C-III, D-IV
(b) A-I, B-II, C-IV, D-III
(c) A-III, B-II, C-IV, D-I
(d) A-II, B-I, C-IV, D-III

166. Which one of the following techniques does not serve the purpose of early diagnosis of a disease for its early treatment?

(a) Recombinant DNA Technology
(b) Serum and Urine analysis
(c) Polymerase Chain Reaction (PCR) technique
(d) Enzyme Linked Immuno-Sorbent Assay (ELISA) technique

167. Match List I with List II.

List I (Interacting species)	**List II (Name of Interaction)**
A. A Leopard and a Lion in a forest/grassland	I. Competition
B. A Cuckoo laying egg in a Crow's nest	II. Brood parasitism
C. Fungi and root of a higher plant in Mycorrtizae	III. Mutualism
D. A cattle egret and a Cattle in a field	IV. Commensalism

Choose the ***correct*** answer from the options given below:

(a) A-I, B-II, C-III, D-IV
(b) A-I, B-II, C-IV, D-III
(c) A-III, B-IV, C-I, D-II
(d) A-II, B-III, C-I, D-IV

168. Given below are two statements:

Statement I: Ligaments are dense irregular tissue.

Statement II: Cartilage is dense regular tissue.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are true.
(b) Both **Statement I** and **Statement II** are false.
(c) **Statement I** is true but **Statement II** is false.
(d) **Statement I** is false but **Statement II** is true.

169. Given below are two statements:

Statement I: In prokaryotes, the positively charged DNA is held with some negatively charged proteins in a region called nucleoid.

Statement II: In eukaryotes, the negatively charged DNA is wrapped around the positively charged histone octamer to form nucleosome.

In the light of the above statements, choose the ***correct*** answer from the options given below:
(a) Both **Statement I** and **Statement II** are true.
(b) Both **Statement I** and **Statement II** are false.
(c) **Statement I** is correct but **Statement II** is false.
(d) **Statement I** incorrect but **Statement II** is true.

170. Match List I with List II with respect to human eye.

List I	**List II**
A. Fovea	I. Visible coloured portion of eye that regulates diameter of pupil
B. Iris	II. External layer of eye formed of dense connective tissue.
C. Blind spot	III. Point of greatest visual acuity or resolution.
D. Sclera	IV. Point where optic nerve leaves the eyeball and photoreceptor cells are absent.

Choose the ***correct*** answer from the options given below:
(a) A-III, B-I, C-IV, D-II
(b) A-IV, B-III, C-II, D-I
(c) A-I, B-IV, C-III, D-II
(d) A-II, B-I, C-III, D-IV

171. Select the ***correct*** group/set of Australian Marsupials exhibiting adaptive radiation.
(a) Tasmanian wolf, Bobcat, Marsupial mole
(b) Numbat, Spotted cuscus, Flying phalanger
(c) Mole, Flying squirrel, Tasmanian tiger cat
(d) Lemur, Anteater, Wolf

172. Which of the following statements are ***correct*** regarding female reproductive cycle?
A. In non-primate mammals cyclical changes during reproduction are called oestrus cycle.
B. First menstrual cycle begins at puberty and is called menopause.
C. Lack of menstruation may be indicative of pregnancy.
D. Cyclic menstruation extends between menarche and menopause.

Choose the **most appropriate** answer from the options given below:
(a) A and D only
(b) A and B only
(c) A, B and C only
(d) A, C and D only

173. Vital capacity of lung is _____
(a) IRV + ERV
(b) IRV + ERV + TV + RV
(c) IRV + ERV + TV – RV
(d) IRV + ERV + TV

174. Match List I with List II.

List I	**List II**
A. P-wave	I. Beginning of systole
B. Q-wave	II. Repolarisation of ventricles
C. QRS complex	III. Depolarisation of atria

D. T-wave IV. Depolarisation of ventricles

Choose the ***correct*** answer from the options given below:

(a) A-III, B-I, C-IV, D-II (b) A-IV, B-III, C-II, D-I

(c) A-II, B-IV, C-I, D-III (d) A-I, B-II, C-III, D-IV

175. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A: Amniocentesis for sex determination is one of the strategies of Reproductive and Child Health Care Programme.

Reason R: Ban on amniocentesis checks increasing menace of female foeticide.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **A** and **R** are true and **R** is the correct explanation of **A**.

(b) Both **A** and **R** are true and **R** is NOT the correct explanation of **A**.

(c) **A** is true but **R** is false.

(d) **A** is false but **R** is true.

176. Once the undigested and unabsorbed substances enter the caecum, their backflow is prevented by

(a) Sphincter of Oddi (b) Ileo - caecal valve

(c) Gastro - oesophageal sphincter (d) Pyloric sphincter

177. Match List I with List II.

List I	**List II**
A. Gene 'a'	I. β -galactosidase
B. Gene 'y'	II. Transacetylase
C. Gene 'i'	III. Permease
D. Gene 'z'	IV. Repressor protein

Choose the ***correct*** answer from the options given below:

(a) A-II, B-I, C-IV, D-III (b) A-II, B-III, C-IV, D-I

(c) A-III, B-IV, C-I, D-II (d) A-III, B-I, C-IV, D-II

178. Match List I with List II.

List I (Cells)	**List II (Secretion)**
A. Peptic cells	I. Mucus
B. Goblet cells	II. Bile juice
C. Oxyntic cells	III. Proenzyme pepsinogen
D. Hepatic cells	IV. HCl and intrinsic factor for absorption of vitamin B_{12}

Choose the ***correct*** answer from the options given below:

(a) A-IV, B-III, C-II, D-I (b) A-II, B-I, C-III, D-IV

(c) A-III, B-I, C-IV, D-II (d) A-II, B-IV, C-I, D-III

179. Which of the following functions is carried out by cytoskeleton in a cell?

(a) Nuclear division (b) Protein synthesis

(c) Motility (d) Transportation

180. Given below are statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A: Nephrons are of two types: Cortical & Juxta medullary, based on their relative position in cortex and medulla.

Reason R: Juxta medullary nephrons have short loop of Henle whereas, cortical nephrons have longer loop of Henle.

In the light of the above statements, choose the ***correct*** answer from the options given below:

(a) Both **A** and **R** are true and **R** is the correct explanation of **A**.

(b) Both **A** and **R** are true but **R** is NOT the correct explanation of **A**.

(c) **A** is true but **R** is false.

(d) **A** is false but **R** is true.

181. Given below are two statements:

Statement I: Electrostatic precipitator is most widely used in thermal power plant.

Statement II: Electrostatic precipitator in thermal power plant removes ionising radiations

In the light of the above statements, choose the ***most appropriate*** answer from the options given below:

(a) Both **Statement I** and **Statement II** are correct.

(b) Both **Statement I** and **Statement II** are incorrect.

(c) **Statement I** is correct but **Statement II** is incorrect.

(d) **Statement I** incorrect but **Statement II** is correct.

182. Broad palm with single palm crease is visible in a person suffering from

(a) Down's syndrome (b) Turner's syndrome

(c) Klinefelter's syndrome (d) Thalassemia

183. Radial symmetry is NOT found in adults of phylum _____.

(a) Ctenophora (b) Hemichordata

(c) Coelenterata (d) Echinodermata

184. In which blood corpuscles, the HIV undergoes replication and produces progeny viruses?

(a) T_H cells (b) B-lymphocytes

(c) Basophils (d) Eosinophils

185. Which of the following is not a cloning vector?

(a) BAC (b) YAC (c) pBR322 (d) Probe

186. Match List I with List II.

List I	**List II**
A. Logistic growth	I. Unlimited resource availability condition
B. Exponential growth	II. Limited resource availability condition
C. Expanding age pyramid	III. The percent individuals of pre-reproductive age is largest followed by reproductive and post reproductive age groups
D. Stable age pyramid	IV. The percent individuals of pre-reproductives and reproductive age group are same

Choose the ***correct*** answer from the options given below:

(a) A-II, B-I, C-III, D-IV (b) A-II, B-III, C-I, D-IV

(c) A-II, B-IV, C-I, D-III (d) A-II, B-IV, C-III, D-I

187. Select the ***correct*** statements with reference to chordates.

A. Presence of a mid-dorsal, solid and double nerve cord.

B. Presence of closed circulatory system.

C. Presence of paired pharyngeal gillslits.

D. Presence of dorsal heart

E. Triploblastic pseudocoelomate animals.

Choose the ***correct*** answer from the options given below:

(a) A, C and D only
(b) B and C only
(c) B, D and E only
(d) C, D and E only

188. The parts of human brain that helps in regulation of sexual behaviour, expression of excitement, pleasure, rage, fear etc. are :
(a) Limbic system and hypothalamus
(b) Corpora quadrigemina and hippocampus
(c) Brain stem and epithalamus
(d) Corpus callosum and thalamus

189. The unique mammalian characteristics are:
(a) hairs, tympanic membrane and mammary glands
(b) hairs, pinna and mammary glands
(c) hairs, pinna and indirect development
(d) pinna, monocondylic skull and mammary glands

190. Which of the following are NOT under the control of thyroid hormone?
A. Maintenance of water and electrolyte balance
B. Regulation of basal metabolic rate
C. Normal rhythm of sleep-wake cycle
D. Development of immune system
E. Support the process of R.B.Cs formation
Choose the ***correct*** answer from the options given below:
(a) A and D only
(b) B and C only
(c) C and D only
(d) D and E only

191. Select the ***correct*** statements.
A. Tetrad formation is seen during Leptotene.
B. During Anaphase, the centromeres split and chromatids separate.
C. Terminalization takes place during Pachytene.
D. Nucleolus, Golgi complex and ER are reformed during Telophase.
E. Crossing over takes place between sister chromatids of homologous chromosome.
Choose the ***correct*** answer from the options given below:
(a) A and C only
(b) B and D only
(c) A, C and E only
(d) B and E only

192. Match List I with List II.

List I	**List II**
A. Mast cells	I. Ciliated epithelium
B. Inner surface of bronchiole	II. Areolar connective tissue
C. Blood	III. Cuboidal epithelium
D. Tubular parts of nephron	IV. Specialised connective tissue

Choose the ***correct*** answer from the options give below:
(a) A-I, B-II, C-IV, D-III
(b) A-II, B-III, C-I, D-IV
(c) A-II, B-I, C-IV, D-III
(d) A-III, B-IV, C-II, D-I

193. Which of the following is characteristic feature of cockroach regarding sexual dimorphism?
(a) Dark brown body colour and anal cerci
(b) Presence of anal styles
(c) Presence of sclerites
(d) Presence of anal cerci

194. Which one of the following is the sequence on corresponding coding strand, if the sequence on mRNA formed is as follows 5'AUCGAUCGAUCGAUCGAUCGAUCGAUCG3'?
(a) 5'UAGCUAGCUAGCUAGCUAGCUAGCUAGC3'
(b) 3'UAGCUAGCUAGCUAGCUAGCUAGCUAGC3'
(c) 5'ATCGATCGATCGATCGATCGATCGATCG3'
(d) 3'ATCGATCGATCGATCGATCGATCGATCG3'

195. In cockroach, excretion is brought about by-
A. Phallic gland
B. Urecose gland
C. Nephrocytes
D. Fat body
E. Collaterial glands
Choose the ***correct*** answer from the options given below:
(a) A and E only
(b) A, B and E only
(c) B, C and D only
(d) B and D only

196. Given below are two statements:
Statement I: During G_0 phase of cell cycle, the cell is metabolically inactive.
Statement II: The centrosome undergoes duplication during S phase of interphase.
In the light of the above statements, choose the ***most appropriate*** answer from the options given below:
(a) Both **Statement I** and **Statement II** are correct.
(b) Both **Statement I** and **Statement II** are incorrect.
(c) **Statement I** is correct but **Statement II** is incorrect.
(d) **Statement I** is incorrect but **Statement II** is correct.

197. Which one of the following is NOT an advantage of inbreeding?
(a) It decreases homozygosity.
(b) It exposes harmful recessive genes that are eliminated by selection.
(c) Elimination of less desirable genes and accumulation of superior genes takes place due to it.
(d) It decreases the productivity of inbred population, after continuous inbreeding.

198. Which of the following statements are ***correct***?
A. An excessive loss of body fluid from the body switches off osmoreceptors.
B. ADH facilitates water reabsorption to prevent diuresis.
C. ANF causes vasodilation.
D. ADH causes increase in blood pressure.
E. ADH is responsible for decrease in GFR.
Choose the ***correct*** answer from the options given below:
(a) A and B only
(b) B, C and D only
(c) A, B and E only
(d) C, D and E only

199. Which of the following statements are ***correct*** regarding skeletal muscle?
A. Muscle bundles are held together by collagenous connective tissue layer called fascicle.
B. Sarcoplasmic reticulum of muscle fibre is a store house of calcium ions.
C. Striated appearance of skeletal muscle fibre is due to distribution pattern of actin and myosin proteins.
D. M line is considered as functional unit of contraction called sarcomere.
Choose the ***most appropriate*** answer from the options given below:
(a) A, B and C only
(b) B and C only
(c) A, C and D only
(d) C and D only

200. Which of the following statements are ***correct***?

A. Basophils are most abundant cells of the total WBCs

B. Basophils secrete histamine, serotonin and heparin

C. Basophils are involved in inflammatory response

D. Basophils have kidney shaped nucleus

E. Basophils are agranulocytes

Choose the ***correct*** answer from the options given below:

(a) D and E only

(b) C and E only

(c) B and C only

(d) A and B only